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Key indicators

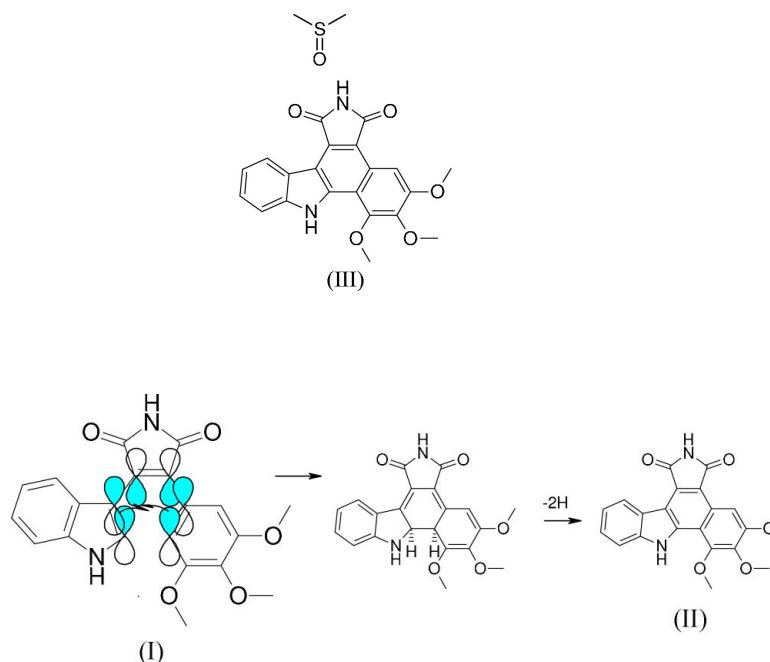
Single-crystal X-ray study
 $T = 295$ K
Mean $\sigma(\text{C}-\text{C}) = 0.005$ Å
 R factor = 0.076
 wR factor = 0.187
Data-to-parameter ratio = 14.2For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.5,6,7-Trimethoxy-2,3-dihydro-1*H*,8*H*-benzo[*a*]-pyrrolo[3,4-*c*]carbazole-1,3-dione dimethyl sulfoxide solvateThe crystal structure of the title compound, $\text{C}_{21}\text{H}_{16}\text{N}_2\text{O}_5 \cdot \text{C}_2\text{H}_6\text{OS}$, was determined to investigate the electrocyclic reactivity of 3,4-diaryl-1*H*-pyrrole-2,5-diones (3,4-bisarylmaleimides) to the yield corresponding carbazole derivatives.

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Comment

The title compound, (III), bearing the carbazole moiety as a core structure, was accidentally isolated from an ethyl acetate solution of 3-(indol-3-yl)-4-(3,4,5-trimethoxyphenyl)-1*H*-pyrrole-2,5-dione, (I) (Peifer *et al.*, 2005) at room temperature. The reaction scheme below shows the disrotatory cyclization of (I) and subsequent oxidation to yield (II).

The analytically pure 1*H*-pyrrole-2,5-dione derivative was found to undergo a reaction (monitored by thin-layer chromatography) producing (II). A comparable mechanism of reactions of the class of 1*H*-pyrrole-2,5-diones had been reported by Sanchez-Martinez *et al.* (2003) and Harris *et al.* (1993). However, after 24 h in an ethyl acetate solution, approximately 10% of (II) could be determined by high-performance liquid chromatographic analysis. Compound (II) was subsequently isolated by column chromatography and found to be chemically stable. Crystals of (I) precipitated at 278 K from DMSO. We now report the X-ray crystal structure analysis of carbazole (III), which is the DMSO solvate of carbazole (II) and which confirms the structure and strongly supports the mechanism of oxidative cyclization of the 1*H*-

pyrrole-2,5-dione derivative to generate compound (II). The solvent DMSO molecule is linked *via* a hydrogen bond to the carbazole molecule (see Table 1 and Fig. 1).

Experimental

The title compound was obtained by crystallization of a DMSO solution of (II).

Crystal data

$C_{21}H_{16}N_2O_5 \cdot C_2H_6OS$

$M_r = 454.50$

Monoclinic, $P2_1/c$

$a = 7.994 (2) \text{ \AA}$

$b = 20.040 (4) \text{ \AA}$

$c = 13.644 (4) \text{ \AA}$

$\beta = 106.586 (12)^\circ$

$V = 2094.8 (9) \text{ \AA}^3$

$Z = 4$

$D_x = 1.441 \text{ Mg m}^{-3}$

Cu $K\alpha$ radiation

Cell parameters from 25

reflections

$\theta = 30\text{--}44^\circ$

$\mu = 1.76 \text{ mm}^{-1}$

$T = 295 (2) \text{ K}$

Needle, yellow

$0.24 \times 0.06 \times 0.04 \text{ mm}$

Data collection

Enraf–Nonius CAD-4
diffractometer

$\omega/2\theta$ scans

Absorption correction: ψ scan
(CORINC; Dräger & Gattow,
1971)

$T_{\min} = 0.783$, $T_{\max} = 0.932$

4534 measured reflections

4228 independent reflections

2943 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.062$

$\theta_{\max} = 74.0^\circ$

$h = -9 \rightarrow 0$

$k = 0 \rightarrow 24$

$l = -16 \rightarrow 17$

3 standard reflections

frequency: 60 min

intensity decay: 5%

Refinement

Refinement on F^2

$R[F^2 > 2\sigma(F^2)] = 0.076$

$wR(F^2) = 0.187$

$S = 1.11$

4228 reflections

298 parameters

H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0644P)^2 + 1.9378P]$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} < 0.001$

$\Delta\rho_{\max} = 0.32 \text{ e \AA}^{-3}$

$\Delta\rho_{\min} = -0.35 \text{ e \AA}^{-3}$

Table 1

Hydrogen-bonding geometry (\AA , $^\circ$).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
$N6-H6 \cdots O21$	0.79	2.20	2.741 (4)	126
$N10-H10 \cdots O2L$	0.87	2.00	2.849 (5)	164

H atoms attached to N were located in a difference map and refined isotropically. Other H atoms were placed at calculated positions and refined with fixed isotropic displacement parameters using a riding model [$C-H = 0.93$ or 0.96 \AA and $U(H) = 1.2$ or 1.5 times $U_{\text{eq}}(C)$]; the methyl groups were allowed to rotate but not to tip.

Data collection: *CAD-4 Software* (Enraf–Nonius, 1989); cell refinement: *CAD-4 Software*; data reduction: *CORINC* (Dräger &

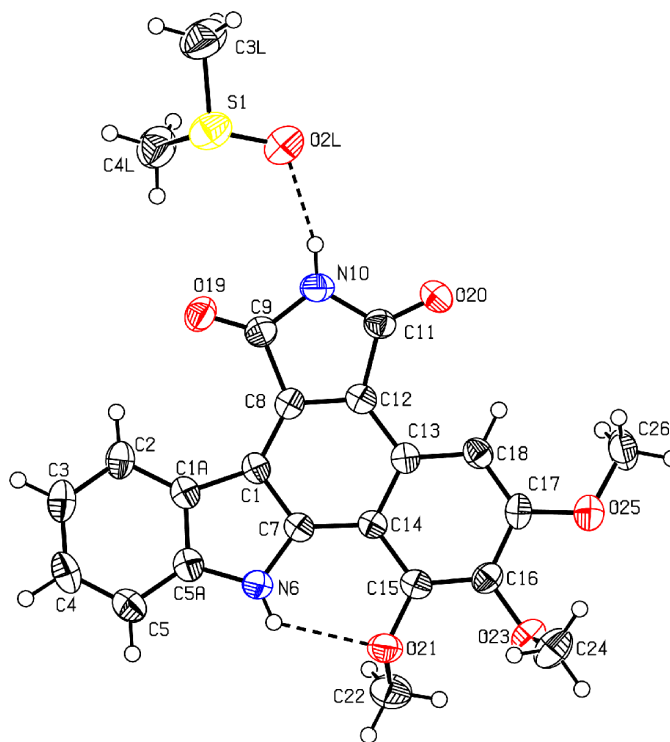


Figure 1

ORTEP view (Johnson, 1976) of (III). Displacement ellipsoids are shown at the 50% probability level. H atoms are depicted as circles of arbitrary size. Dashed lines indicate hydrogen bonds.

Gattow, 1971); program(s) used to solve structure: *SIR92* (Altomare *et al.*, 1994); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *PLATON* (Spek, 2003); software used to prepare material for publication: *SHELXL97*.

References

- Altomare, A., Cascarano, G., Giacovazzo, C., Guagliardi, A., Burla, M. C., Polidori, G. & Camalli, M. (1994). *J. Appl. Cryst.* **27**, 435.
- Dräger, M. & Gattow, G. (1971). *Acta Chem. Scand.* **25**, 761–762.
- Enraf–Nonius (1989). *CAD-4 Software*. Version 5. Enraf–Nonius, Delft, The Netherlands.
- Harris, W., Hill, C. H., Keech, E. & Malsher, P. (1993). *Tetrahedron Lett.* **34**, 8361–8364.
- Johnson, C. K. (1976). *ORTEP*. ORNL-5138, revised. Oak Ridge National Laboratory, Tennessee, USA.
- Peifer, C., Schollmeyer, D. & Dannhardt, G. (2005). *Acta Cryst.* **E61**. Submitted. **[BT6600]**
- Sanchez-Martinez, C., Faul, M. M., Shih, C., Sullivan, K. A., Grutsch, J. L., Cooper, J. T. & Kolis, S. P. (2003). *J. Org. Chem.* **68**, 8008–8014.
- Sheldrick, G. M. (1997). *SHELXS97*. University of Göttingen, Germany.
- Spek, A. L. (2003). *J. Appl. Cryst.* **36**, 7–13.